A FACILE SYNTHESIS OF 6-OXO-PGF $_{1\alpha}$  AND 6-OXO-PGE $_{1}^{1)}$ 

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6-Oxo-PGF<sub>1α</sub> and 6-oxo-PGE<sub>1</sub> were synthesized from 7-t-butyldimethylsililoxy-6-[(lE)-3-t-butyldimethylsililoxy-1-octenyl]-2-oxabicyclo[3.3.0]octan-2-one ( $\underline{1}$ ) and 7-tetrahydropyranyloxy-6-[(lE)-3-tetrahydropyranyloxy-1-octenyl]-2-oxabicyclo[3.3.0]octan-2-one ( $\underline{1}$ ') via C<sub>5</sub> unit elongation to the lactone carbonyl of  $\underline{1}$  and  $\underline{1}$ ' respectively. The C<sub>5</sub> unit was introduced step by step (C<sub>1</sub> + C<sub>4</sub>) by means of the conversion of  $\underline{1}$  and  $\underline{1}$ ' to the corresponding β-keto esters and then the alkylation of the active methylene of them.

6-Oxo-PGE $_1$  ( $\underline{7}$ ) is one of the most attractive target in prostaglandin (PG) synthesis, because of its higher vaso-activities and availabilities. There are a few reports on the synthetic study, e.g., syntheses of 6-oxo-PGF $_{1\alpha}$  ( $\underline{6}$ ) and  $\underline{7}$  via the 6-nitro-PGE $_1$  derivative by Noyori et al., conversion of PGF $_{2\alpha}$  to  $\underline{6}$  by Johnson et al., and conversion of 6-oxo-PGF $_{1\alpha}$  11,15-ditetrahydropyranyl ether ( $\underline{5}$ ', R=H) to 7 by Nicolaou et al.

Our strategy for the synthesis of 6-oxo-PGs is to build the 6-oxo ester functionality by joining the  $C_5$  unit ( $C^1-C^5$  part, PG numbering) to lactones  $\underline{1}$  and  $\underline{1}$ . Because, the lactones are easily available as an important intermediate for PGF<sub>20</sub> synthesis. However, it is well known that there is no suitable method to

COOMe

R'O

OR'

$$1, 1'$$

OR'

 $1, 1'$ 

OR'

OR'

OR'

OR'

 $\underline{1}$ ,  $\underline{2}$ , and  $\underline{5}$ ;  $R'=R=SiMe_2Bu^t$ ,  $\underline{1}'$ ,  $\underline{2}'$ , and  $\underline{5}'$ ; R'=THP,  $R=SiMe_2Bu^t$ Scheme 1.

 $6\text{-}0\text{xo-PGF}_{1\alpha}$  was synthesized as follows. Lactone  $\underline{1}$  was hydrolyzed with alcoholic potassium hydroxide, and the resulting salt was washed with hexane and dried well under reduced pressure to give a white powder. To a stirred solution of the salt in dimethylformamide (DMF) was added iodomethane (1.5 equiv.), and the stirring was continued for 1 h at r.t. Imidazole (4 equiv.) and t-butyldimethyl-chlorosilane (2 equiv.) were successively added to the reaction mixture, and the stirring was continued for additional 10 h to give methyl ester  $\underline{2}$  (90% yield after column chromatography on silica gel). The ester  $\underline{2}$  was converted to ketones  $\underline{3a}$ - $\underline{d}$ , which have an electron-withdrowing group (EWG) attaching active methylene group, by the reaction with the corresponding carbanion as shown in Table 1.

Table l.	Reaction of	Ester	2	(1	mmol)	with	Carbanion	to	Ketone	3

Formation of Carbanion					Rea	ction wi	Product 3		
CH <sub>3</sub> -EWG (mmol)	LDA (mmol)	Temp	Time h	Solvent (ml)	Method <sup>a</sup>	Temp °C	Time h	Yield/ EWG=	, <sub>&amp;</sub> b)
CH <sub>3</sub> SCH <sub>3</sub> (12	2.5	0	0.5	THF (5)	A	0	1.0	-SCH <sub>3</sub>	86
PhSCH <sub>2</sub> COOH		0	1.0	THF (4)	A	0-r.t.	2.5	-SPh <u>3b</u>	71
СH <sub>3</sub> СООВu <sup>t</sup> (1.		-40	0.5	THF-hexane (2) (2)	В	30	1.5	-соови <sup>t</sup> <u>Зс</u>	:
					В	-40	1.5		27 <sup>c)</sup> 61 <sup>d)</sup>
					В	0	0.5	3c 3c	61 <sup>d)</sup>
					В	r.t.	0.5	<u>3c</u>	67 <sup>e)</sup>
CH <sub>3</sub> COOEt (1.	2.5	-40	0.5	THF-hexane (2) (2)	В	30	0.5	-COOEt	81 <sup>f)</sup>

a) A: Ester 2 was added to the carbanion.

Alkylation of ketones  $\underline{3a}$ - $\underline{d}$  to  $\underline{4a}$ - $\underline{d}$  was carried out by treatment with base and methyl 4-iodobutyrate as shown in Table 2. 6-Oxo-PGF<sub>1 $\alpha$ </sub> tri-t-butyldimethylsilil

B: The carbanion was added to a solution of ester 2 in THF (1 ml).

b) Isolated yield based on 2.

c) Alcohol ( $\frac{5}{2}$ C(OH)(CH<sub>2</sub>COOBu<sup>t</sup>)<sub>2</sub>, 5%) and  $\frac{2}{2}$  (65%) were isolated.

d) Ester 2 (27%) was recovered.

e)  $\beta$ ,  $\delta$ -Diketo ester ( $\frac{1}{2}$ COCH<sub>2</sub>COCH<sub>2</sub>COOBu<sup>t</sup>, 3%) and  $\frac{2}{2}$  (23%) were isolated.

f) Ester 2 (13%) was recovered.

RO EWG COOMe 
$$\frac{3}{3}$$
 and  $\frac{4}{4}$  R'=R=SiMe<sub>2</sub>Bu<sup>t</sup>

R'O  $\frac{3}{3}$ ,  $\frac{3}{3}$ ' OR'  $\frac{4}{4}$ ,  $\frac{4}{4}$ ' OR'  $\frac{3}{8}$ '=THP, R=SiMe<sub>2</sub>Bu<sup>t</sup>

ether  $(\underline{5})$  was obtained by desulfurization of  $\underline{4a}$  (70% yield, excess Al(Hg)) or deethoxycarbonylation of  $\underline{4d}$  (93% yield, alkaline hydrolysis, decarboxylation, and esterification (diazomethane)), though the conversion of  $\underline{4b}$  and  $\underline{4c}$  to  $\underline{5}$  was not successful by the usual manner. Deprotection of  $\underline{5}$  to  $\underline{6}$  was achieved by treatment with tetrabutylammonium fluoride in dry THF at r.t. (90% yield).

Table 2. Alkylation of Ketones 3 to 4

3 EWG=	Base Solvent (mol equiv.)		Temp Time h		4 Yield/%		
3a -SCH <sub>3</sub>	к <sub>2</sub> со <sub>3</sub> (2)	DMF	r.t.	48	<u>4a</u>	65	
<u>3b</u> -SPh	NaH (1.1)	DMSO-DME	0	1	<u>4b</u>	66	
<u>3c</u> -COOBu <sup>t</sup>	<sup>K</sup> 2 <sup>CO</sup> 3 (2)	DME	30	72	<u>4c</u>	62 <sup>a)</sup>	
<u>3d</u> -COOEt	к <sub>2</sub> со <sub>3</sub> (2)	DME	30	48	<u>4d</u>	93 <sup>b)</sup>	

a) Ketone 3c (36%) was recovered.

b) Methyl 4-iodobutyrate was used in excess (ca. 2 equiv.).

hydrolyzed to the corresponding dicarboxylic acid by treatment with alcoholic sodium hydroxide and followed with diluted hydrochloric acid. The crude acid was heated with NaHCO $_3$  (ca. 0.5 mol equiv. relative to the carboxylic acid) in refluxing benzene for 1 h. The resulting 4-oxo carboxylic acid was treated with etheral diazomethane to give the methyl ester  $\underline{5}$ ' in 93% yield based on  $\underline{4d}$ ' after column chromatography on silica gel. Desililation of  $\underline{5}$ ' to 6-oxo-PGF $_{1\alpha}$  11,15-ditetrahydropyranyl ether, which was converted to 6-oxo-PGE $_1$  ( $\underline{7}$ ) by oxidation followed by deprotection, was achieved by treatment with tetrabutylammonium fluoride in THF in 90% yield.

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- 7) There is no report on the preparation of  $\beta$ -keto ester from ester by treatment with ester enolate, though there are several reports on that from acid chlorides or acid anhydrides with ester enolates (W. Wierenga and H. I. Skulnick, J. Org. Chem.,  $\underline{44}$ , 310 (1979); see reviews, N. Petragnani and M. Yonashiro, Synthesis, 1982, 521).
- 8) We could not obtain satisfactory results by reductive desulfurization of  $\underline{4b}$  (Raney nickel, Al(Hg), or PhSNa) and by acid catalyzed hydrolysis of  $\underline{4c}$ .

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